

Heat Pulse Charging: Nonlinear Regression Fit

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December 2008

Line Charge Thermal Model (Bristow 94, page 449)

$$T(t) = T_0 + \frac{-Q'}{4\pi\kappa} \text{Ei} \left(\frac{-r^2}{4\kappa t} \right) \quad (1)$$

Q' : is the source strength per unit time ($m^2 \text{ } ^\circ\text{C}s^{-1}$)

κ : thermal diffusivity m^2s^{-1}

r : 'effective' radius (meters)

Note: $\text{Ei}(-x) = -\text{E}_1(x)$

Ball park figures

$(Q' , \kappa, r) = [1.287\text{e-}5 \text{ } m^2\text{degC}s^{-1}, 5\text{e-}7 \text{ } m^2s^{-1}, 1\text{e-}3 \text{ meters}]$

Nonlinear fit to parameters: Fig 1 upper curve

Need starting values and bounds (min, max) for Q' , κ , and r ?

Starting values $(Q', \kappa, r) = [1.287e-5, 5e-7, 1e-4]$ – Q is set fixed to estimate.

Here is the fit for approx $(Q', \kappa, r) =$

$(0.000012870000000 0.000000227403477 0.001213448506939)$:

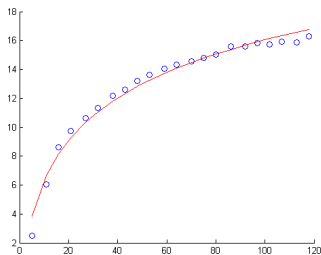


Figure: Fit with $Q = 1.287e-5$, Upper

So 'r' (effective radius) is approximately 1 mm.

Nonlinear fit to parameters: Lower curve

Starting values (Q' , κ , r): [1.083e-5, 5e-7, 1e-4]

$(Q', \kappa, r) = (1.0e-003) * (0.010830000000000000 \ 0.000310406551974 \ 0.952919764737804)$

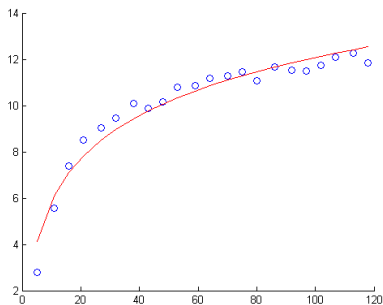


Figure: Fit with $Q = 1.083e-5$, Lower

So 'r' is approximately 1 mm again.

Method used to estimate Q'

Well if we know the soil moisture content (θ) at any location, we can estimate ρc as follow:

$$\rho c = (\rho c)_s + (\rho c)_w \theta_s$$

where $(\rho c)_s$ = volumetric heat capacity for soil

$(\rho c)_w$ = volumetric heat capacity for water

$\rho_{\text{sand}} = 1.62 \text{ g/cm}^3$ (measured in the sand column)

c_s the specific heat capacity of soil solid = $0.83 \text{ J g}^{-1} \text{ }^\circ\text{C}^{-1}$ (estimated for quartz sand)

c_w the specific heat capacity of water = $4.186 \text{ J g}^{-1} \text{ }^\circ\text{C}^{-1}$ (at $15 \text{ }^\circ\text{C}$)

$\rho_w = 9991 \text{ g/cm}^3$ (at $15 \text{ }^\circ\text{C}$)

If you plug in the water content for the 2 datasets you have you will get:

$$\rho c = 1.554 \text{ MJ/m}^3 \text{ }^\circ\text{C for } \theta = 0.05 \text{ m}^3/\text{m}^3$$

$$\text{and } \rho c = 1.846 \text{ MJ/m}^3 \text{ }^\circ\text{C for } \theta = 0.12 \text{ m}^3/\text{m}^3$$

Knowing that $Q' = q/\rho c$ and $q = 20 \text{ J m}^{-1} \text{ s}^{-1}$ we can calculate Q' for each water content.

Thus, we can use the calculated Q' to find r for the two datasets and see if it is changing with moisture content.

For the thermal diffusivity, my best estimate is that it can vary from $1 \times 10^{-7} \text{ m}^2 \text{ s}^{-1}$ to $1 \times 10^{-6} \text{ m}^2 \text{ s}^{-1}$

Q' estimate

Chadi Paper, Figure 1, 'Upper Curve':

$$\text{For } \theta = 0.05 \text{ m}^3/\text{m}^3 \text{ --- } > Q=20/1.544\text{e}6 = 1.287\text{e-}5 \text{ m}^2 \text{ C s-1}$$

Chadi Paper, Figure 1, 'Lower Curve':

$$\text{For } \theta = 0.12 \text{ m}^3/\text{m}^3 \text{ --- } > Q=20/1.846\text{e}6 = 1.083\text{e-}5 \text{ m}^2 \text{ C s-1}$$

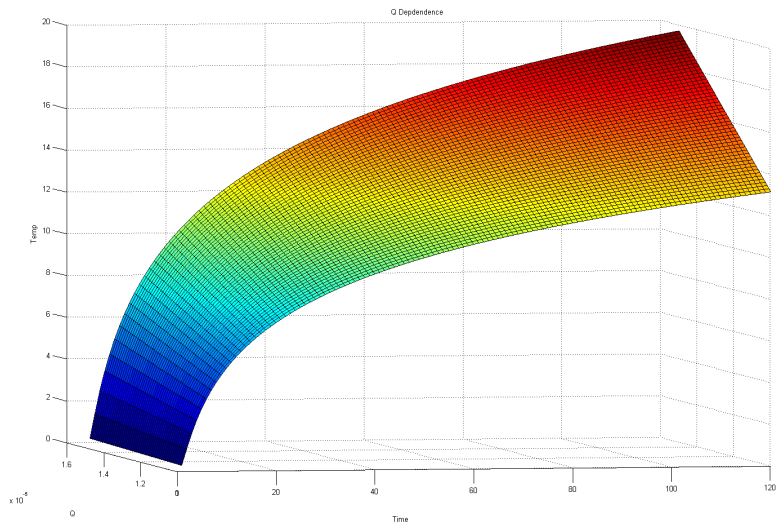
Thermal Integral Method Solution

If time of heat pulse is 'fixed', call it t_f , than the 'thermal integral' can be approximated by

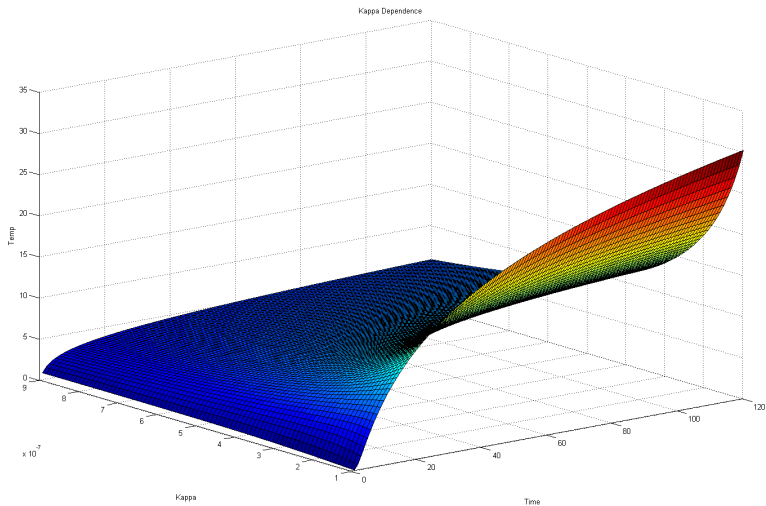
$$\int_0^{t_f} \Delta T dt = \frac{Q'}{4\pi\kappa} [(t_f - t_c) \cdot \ln(t_f - t_c) + d \cdot t_f + A]$$

for constants t_c , d , and A .

Q Dependence



Kappa Dependence



Radius Dependence

